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pg. 24 Selected Honour Chemistry Assignment Answers Unit 6: Acids and Bases Chapter 16: Acids and Bases (pg. 568 to 571) 1. Brnsted acids are chemicals that can donate protons, whereas Brnsted bases are chemicals that can accept protons. In Brnsted definitions of acids and bases, water is an active participant during the dissociation process. When Brnsted acid dissociates, water acts like a base to a...

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